

# New Smells, New Ideas Ball-and-Stick Models

Name \_\_\_\_\_ Period \_\_\_\_\_

**Purpose** 

To examine ball-and-stick models of molecules and compare them to structural formulas.

## Part I: New Smell Molecules

Test the samples for smell. Fill in the table, then answer the questions.

Vial	Molecular formula and name	Functional group	Compound type	Structural formula	Smell
Ι	C <sub>10</sub> H <sub>20</sub> O citronellol			$ \begin{array}{c} H \\ -O \\ H \\ -C \\ -H \\ -H \\ H \\ -C \\ -H \\ -H$	
J	C <sub>10</sub> H <sub>18</sub> O fenchol			H = H = H = H	



### Questions

- I. Circle and identify the functional group in each structural formula.
- **2.** Can functional group alone be used to classify these five molecules according to their smell? Why or why not?

- **3.** Is the molecular formula alone enough information to allow these five molecules to be classified according to their smell? Why or why not?
- **4.** Are there any structural similarities besides functional group that might be used to classify these molecules? If so, what are they?

#### Part 2: Three-Dimensional Models

Examine the three ball-and-stick models. They represent the molecular compounds you smelled today. Figure out the molecular formula, name, and smell of each and write them in the table.

	Molecule I	Molecule 2	Molecule 3
Molecular formula			
Name			
Smell			



#### Questions

- I. Compare the three models. List at least three physical differences that you notice between them.
- **2. Making Sense** How is a ball-and-stick representation different from a structural formula? What additional information does it convey?
- **3. If You Finish Early** Exactly what molecular model pieces would you need in order to construct a ball-and-stick molecular model of menthol?