

## **Beyond What You See Electromagnetic Radiation**

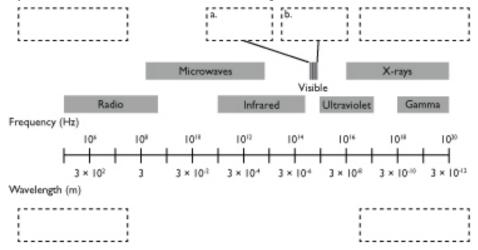
Name _	
Date	Period

Purpose

To explore light of wavelengths and frequencies that humans cannot see.

## Part I: Iron Lung

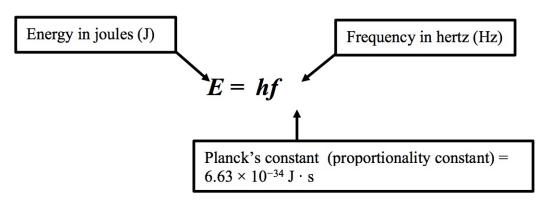
The light that we see is part of a larger collection of waves called *electromagnetic radiation*. The wavelength, frequency, and energy of electromagnetic radiation vary widely, but these waves all travel with the same speed.



- I. In the empty boxes on the left and right of the diagram above, label the radiation with the highest frequency, lowest frequency, longest wavelength, and shortest wavelength.
- **2.** Visible radiation (the rainbow) is spread between the boxes labeled (a) and (b). In the boxes provided, indicate the color you would see nearest to (a) and nearest to (b).

## Part 2: Light Energy

The diagram above shows waves that represent red, yellow, green, and purple light. The distance between peaks is called the wavelength,  $\lambda$  (pronounced lambda). The number of waves that pass by per second is called the frequency, *f*.



**I.** Fill in the missing blanks in the table.

Type of radiation	Wavelength, $\lambda$ (m)	Frequency, <i>f</i> (Hz)	Planck's constant, <i>h</i> (J · s)	Energy, E (J)
Gamma rays	$3 \times 10^{-12} \mathrm{m}$	10 <sup>20</sup> Hz	$6.63 \times 10^{-34} \text{ J} \cdot \text{s}$	$6.63 \times 10^{-14} \text{ J}$
X-rays		10 <sup>18</sup> Hz		6.63 × 10 <sup>-16</sup> J
Ultraviolet	$3 \times 10^{-8} \mathrm{m}$	10 <sup>16</sup> Hz	$6.63 \times 10^{-34} \text{ J} \cdot \text{s}$	
Visible	6 × 10 <sup>-7</sup> m	$5 \times 10^{14} \text{ Hz}$	$6.63 \times 10^{-34} \text{ J} \cdot \text{s}$	
Infrared	3 × 10 <sup>-5</sup> m		$6.63 \times 10^{-34} \text{ J} \cdot \text{s}$	$6.63 \times 10^{-21} \text{ J}$
Microwaves		10 <sup>10</sup> Hz		$6.63 \times 10^{-24} \text{ J}$
Radio waves	3 × 10 <sup>3</sup> m		$6.63 \times 10^{-34} \text{ J} \cdot \text{s}$	$6.63 \times 10^{-29} \text{ J}$

- **2.** Which type of radiation is considered the most harmful because it has the highest energy?
- **3.** Which type of radiation is considered the least harmful because it has the lowest energy?
- **4.** There are two equations that describe electromagnetic radiation.

$$\lambda \times f = c$$
$$E = h \times f$$

Write an equation that relates the energy to the wavelength associated with that type of radiation. Hint: Determine the two variables that both equations above have in common. Solve for that variable in one equation, and substitute it into the other.

**5.** Describe in your own words how the frequency of an electromagnetic wave is related to the energy associated with the wave.

**6.** Describe in your own words how the wavelength of electromagnetic radiation is related to the energy of the wave.

**7. Making Sense** Describe in your own words how all types of electromagnetic radiation are similar and how they are different.