esson 9

**ACTIVITY** 

# Create a Table Properties of the Elements

Name	
Date	Period

## **Purpose**

To create your own periodic table of the elements from data given on element cards.

#### **Materials**

Create a Table card deck

#### **Instructions**

- 1. Work in your groups with one set of cards.
- **2.** Find Be, Mg, Ca, and Sr in the deck of cards, and arrange them in a column the way Mendeleyev did. These cards are all yellow. Look for similarities and differences in these cards. Find at least one pattern or trend, and describe it to your group.
- **3.** With your team, decide how to organize the rest of the cards into a table. Try to organize them in a way that produces as many patterns as possible.

### **Questions**

- 1. What characteristics did you use to sort the cards?
- **2.** What patterns appear in your arrangement? List at least four.
- reacts only slightly found in with water solid MgCl.

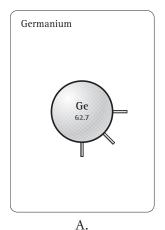
  Calcium moderately hard, silvery solid metal

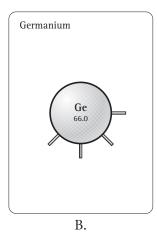
  Ca 40.1

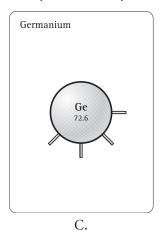
  reacts found in solid CaCl.

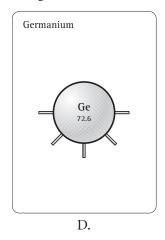
  Strontium moderately soft, silvery white solid metal
- **3.** Where did you put H and He? What was your reasoning for their placement?
- **4.** Did you notice any cards that didn't quite fit or that seemed out of order? Explain.

- **5. Making Sense** Below are four possible cards for the element germanium, Ge.
  - **a.** Where does germanium belong in the table?
  - **b.** Which card seems most accurate to you? What is your reasoning?

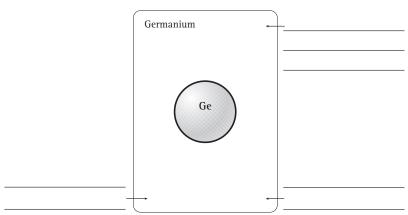








**c.** Copy your selection from part b. What would you add to the three empty corners to complete the card?



**6. If You Finish Early** The element cesium, Cs, is located just below rubidium, Rb, on the modern periodic table. Create an element card for cesium.